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## Physical Science Study Guide

## Multiple Choice

Identify the choice that best completes the statement or answers the question.
$\qquad$ 1. The electrons in a water molecule are gathered nearest to
a. the one oxygen atom.
c. the four hydrogen atoms.
b. the two hydrogen atoms.
d. None of the above
$\qquad$ 2. Compounds that do not dissolve in water are usually
a. polar.
c. homogenous.
b. dispersed.
d. nonpolar.
$\qquad$ 3. Loose sugar dissolves much faster than a sugar cube because loose sugar
a. has a greater surface area.
c. has a higher temperature.
b. has less kinetic energy.
d. has a greater surface tension.
$\qquad$ 4. Molarity is a measure of concentration based on moles of
a. solute per liter of solution.
c. solute per 100 g of solution.
b. solvent per liter of solution.
d. solvent per 100 g of solution.
$\qquad$ 5. What is a salt?
a. a solution that contains a strong acid
b. an ionic compound that does not contain oxide or hydroxide anions
c. a substance formed by mixing two strong acids together
d. a solution that contains more anions than cations and is strongly acidic
6. Yogurt is made by changing
a. milk fat into an acid solution.
b. acid in milk into a strong base.
c. lactose (milk sugar) into lactic acid.
d. protein in milk into a denatured solid.
$\qquad$ 7. A base forms which ions in solution?
a. oxygen
c. hydronium
b. hydroxide
d. hydrogen
8. Which phrase is not true about strong acids?
a. conduct electricity well
c. turn red litmus paper blue
b. ionize completely in water
d. form $\mathrm{H}_{3} \mathrm{O}^{+}$ions in solution
9. What does the pH of a solution measure?
a. concentration of $\mathrm{H}_{3} \mathrm{O}^{+}$ions
c. concentration of metal ions
b. molarity of the solute
d. solubility of the solute
10. A 0.00001 M solution of the strong acid HCl has a pH of
a. 2
b. 3
c. 4
d. 5
11. Soaps and detergents are
a. molecular compounds.
c. ionic compounds.
b. nonpolar compounds.
d. covalent compounds.
12. When a solution of an acid reacts with a solution of a base, hydronium ions react with hydroxide ions to form
a. salt.
c. a weaker base.
b. a stronger acid.
d. water.
13. Which solution is the most acidic?
a. a solution with a $\mathrm{pH}=10$
c. a solution with a $\mathrm{pH}=3$
b. a solution with a $\mathrm{pH}=6$
d. a solution with a $\mathrm{pH}=1$
14. A sample of plutonium-239 decays to one-eighth of its original amount after $7.236 \times 10^{4}$ years. What is its half-life?
a. $\quad 2.412 \times 10^{4}$ years
b. $3.618 \times 10^{4}$ years
c. $7.236 \times 10^{4}$ years
d. $1.447 \times 10^{5}$ years
15. In a stable nucleus, the attractive forces are $\qquad$ the repulsive forces.
a. weaker than
c. canceled out by
b. stronger than
d. equal to
16. The process of nuclear change in an atom of radioactive material is called
a. radioactive decay.
c. nuclear mass.
b. isotopes.
d. radon.
17. The opposite of fusion is called
a. beta decay.
c. fission.
b. alpha decay.
d. neutron transmission.
18. You prepare a large screened-in box, inside of which you place several dozen mouse traps. You set each trap, and on each mouse trap you place a table tennis ball. You then drop another ball into the box, which sets off one of the mouse traps, which sets off other mouse traps, and so on. You have just demonstrated
a. a chain reaction.
c. fusion.
b. the theory of relativity.
d. alpha decay.
19. On a speed-time graph, a line with a negative slope indicates that the object is
a. speeding up.
c. not moving.
b. slowing down.
d. traveling at a constant speed.
20. An object is in motion when
a. you observe the object move.
b. the object's speed increases.
c. the object's displacement is greater than the distance traveled.
d. the object changes position relative to a stationary reference point.
21. When the motion of an object is shown by a straight line on a distance vs. time graph with distance on the $y$-axis, the slope of the line is the
a. distance traveled.
c. speed.
b. displacement.
d. time of travel.
22. The distance traveled by an object divided by the time it takes to travel that distance is called
a. average velocity.
c. average acceleration.
b. average speed.
d. negative acceleration.
23. An object is in motion when
a. the net force acting on the object is zero.
b. friction occurs.
c. the object's displacement is greater than the distance traveled.
d. the object changes position relative to a frame of reference.
24. Acceleration is defined as the change in velocity divided by
a. speed.
c. time.
b. final velocity.
d. distance.
25. Which statement about weight is incorrect?
a. An object weighs more on the moon than it weighs on Earth.
b. A change in an object's location can change the object's weight.
c. An object's weight is directly proportional to its mass.
d. The weight of an object depends on gravity.
26. Which statement about action-reaction force pairs is incorrect?
a. They act on the same object.
c. They occur at the same time.
b. They always occur in pairs.
d. They are equal and opposite.
27. Two identical cars are traveling at equal speeds. One is going north and the other south. If they collide head-on and stick together, what will happen to the motion of the cars?
a. Both cars will move north.
c. Both cars will move south.
b. Both cars will move east.
d. Both cars will stop.
28. When a moving bowling ball hits a pin, some of the ball's momentum
a. doubles in force.
c. is transferred to the pin.
b. increases the pin's mass.
d. is lost.
29. Which of the following is not a factor in calculating momentum?
a. mass
c. acceleration
b. direction
d. speed
30. Whenever an object is standing still, which value is always zero?
a. speed
c. momentum
b. velocity
d. All of the above
31. Near Earth's surface, an object's free-fall acceleration increases as its
a. mass increases.
c. speed increases.
b. weight increases.
d. None of the above
32. What is the kinetic energy of a 1.40 kg discus with a speed of $22.5 \mathrm{~m} / \mathrm{s}$ ?
a. $\quad 15.8 \mathrm{~J}$
b. $\quad 31.5 \mathrm{~J}$
c. 354 J
d. 709 J
33. Mechanical energy can change to nonmechanical energy as a result of
a. air resistance.
c. radiation.
b. heat.
d. None of the above
34. An $84 \%$ efficient single pulley is used to lift a 230 kg piano 3.5 m . How much work must be input?
a. 676 J
b. $9.6 \times 10^{2} \mathrm{~J}$
c. $\quad 6.6 \times 10^{3} \mathrm{~J}$
d. $\quad 9.4 \times 10^{3} \mathrm{~J}$
35. An inclined plane
a. changes the direction of the force only.
b. changes the magnitude of the force only.
c. changes both the magnitude and the direction of the force.
d. decreases the amount of work done.
36. What is the gravitational potential energy of a 54 kg box that is 8.0 m above the ground?
a. $\quad 5,500 \mathrm{~J}$
b. $4,300 \mathrm{~J}$
c. $3,400 \mathrm{~J}$
d. 550 J
37. A man pushes a crate along a factory floor by exerting a force of 55 N . If the crate moves a distance of 4.0 m , how much work does the man perform?
a. $\quad 165 \mathrm{~N}$
c. zero
b. 220 N
d. 145 J
38. What are the units of power?
a. watts
c. joules per second
b. horsepower
d. All of the above
39. Which of the following is not in the inclined plane family?
a. a wedge
c. a ramp
b. a screw
d. a wheel and axle
40. Which of the following statements about work and energy is not true?
a. When work is done, energy is transferred or transformed.
b. Energy may be defined as the ability to do work.
c. Work and energy are always equal.
d. Work and energy have the same units.
41. The kind of energy associated with atomic bonds is
a. nuclear energy.
c. chemical energy.
b. light energy.
d. kinetic energy.
42. When energy is transformed, the amount of usable energy
a. decreases.
c. increases.
b. remains constant.
d. None of the above
43. Using the following table, determine which substance can absorb the most energy in a temperature increase of 1 K .

## SPECIFIC HEATS AT $\mathbf{2 5}^{\circ} \mathbf{C}$

| Substance | $\boldsymbol{c}(\mathbf{J} / \mathbf{k g} \cdot \mathbf{K})$ | Substance | $\boldsymbol{c}(\mathbf{J} / \mathbf{k g} \cdot \mathbf{K})$ |
| :--- | :---: | :--- | :---: |
| Water (liquid) | 4,186 | Copper | 385 |
| Steam | 1,870 | Gold | 129 |
| Ammonia (gas) | 2,060 | Iron | 449 |
| Ethanol (liquid) | 2,440 | Mercury | 140 |
| Aluminum | 897 | Lead | 129 |
| Carbon (graphite) | 709 | Silver | 234 |

a. liquid water
c. gold
b. aluminum
d. lead
44. How much heat energy will cause the temperature of 7.0 kg of iron to increase its temperature by 15 K ? The specific heat of iron is $449 \mathrm{~J} / \mathrm{kg} \bullet \mathrm{K}$.
a. $\quad 6.8 \times 10^{4} \mathrm{~J}$
b. $\quad 4.7 \times 10^{4} \mathrm{~J}$
c. $\quad 7.0 \times 10^{4} \mathrm{~J}$
d. $\quad 3.0 \times 10^{4} \mathrm{~J}$
45. Sound waves from a radio generally travel in which medium?
a. air
c. light
b. earth
d. water
46. If a transverse wave is moving from right to left, the individual particles in the medium are moving
a. right to left.
c. up and down.
b. left to right.
d. None of the above
$\qquad$ 47. The $\qquad$ is the time it takes for one full vibration of a particle in a medium.
a. period
c. amplitude
b. frequency
d. wave speed
48. The speed of a sound wave
a. depends on wavelength.
c. depends on amplitude.
b. depends on the medium.
d. None of the above
49. The frequency of a sound wave determines
a. the pitch of the sound.
c. how fast the sound travels.
b. how loud the sound is.
d. the magnitude of the compression.
50. The color of light is determined by the $\qquad$ of the light waves.
a. medium
c. frequency
b. speed
d. amplitude

